

# ELECTROLYSIS OF MOLTEN IONIC COMPOUNDS

Sr. no.	Ionic compound	+ive ions formed	-ive ions formed	Product formed at the +ive electrode	Product formed at the -ive electrode	Half equation for +ive electrode	Half equations for -ive electrode	Marks
1.	NaCl							6
2.	CaBr <sub>2</sub>							6
3.	CuCl <sub>2</sub>							6
4.	FeCl <sub>3</sub>							6
5.	NiCl <sub>2</sub>							6
6.	Li <sub>2</sub> S							6
7.	AgBr							6

Total marks (42)