

THE MOLE & CONCENTRATION MARK SCHEME

Q1.

Question	Answer	Extra information	Marks
	0.06	correct answer with or without working = 2 marks if answer is incorrect $(0.1 \times 15)/25$ or 0.0015×40 gains 1 mark	2
Total marks			2

Q2.

Question	Answer	Extra information	Marks
	17.6/44 (moles) or 0.4 (moles) CO ₂		1
	7.2/18 (moles) or 0.4 (moles) H ₂ O		1
	empirical formula = CH ₂	allow 1C:2H or correct simplest ratio related to elements or ecf from previous stage allow this mark for correct formula alone	1
Total marks			3

Q3.

Question	Answer	Extra information	Marks
(a)	reasonable smooth curve through all the points over the range 10 - 80	ignore outside range do not accept multiple lines	1
(b)	5.7	range 5.5–5.9 if outside range check graph	1
(c)	7.6	correct answer with or without working = 2 marks if answer incorrect 10 or 2.4 gains 1 mark	2
Total marks			4

Q4.

Question	Answer	Extra information	Marks
	0.11(04)	correct answer with or without working = 2 marks if answer incorrect $(0.15 \times 18.4) / 25$ gains 1 mark	2
Total marks			2

Q5.

Question	Answer	Extra information	Marks
	$\begin{array}{l} \text{CO}_2 \quad 2\text{H}_2\text{O} \\ \frac{1.1}{44} \quad \frac{0.9}{18} \\ = 0.025 \quad = 0.05 \\ 1 \text{ (mole) CO}_2 \quad 2 \text{ (moles) H}_2\text{O} \\ 1 \quad 4 \\ \text{or} \\ \text{CH}_4 \\ \text{or alternative method} \\ \text{Mass of C} = \frac{12}{44} \times 1.1 = 0.3\text{g} \quad (1) \\ \text{Mass of H} = \frac{2}{18} \times 0.9 = 0.1\text{g} \quad (1) \\ \text{C : H} \\ \frac{\text{mass}}{M_r} \quad \frac{0.3}{12} : \frac{0.1}{1} \\ \text{proportions } 0.025 : 0.1 \quad (1) \\ \text{whole number } 1 : 4 \quad (1) \\ \text{or} \\ \text{CH}_4 \end{array}$	<p>correct formula with no working is only 1 mark M3 can be awarded from the formula if steps one and two are clear correct formula from their incorrect ratio gets 1 mark if fraction is wrong way around e.g. M_r / mass, then lose M1 and M2 but accept ecf for M3 and M4</p>	<p>1 1 1 1</p>
Total marks			4

Q6.

Question	Answer	Extra information	Marks
(i)	0.2255 / 0.226 / 0.23 (mole)	correct answer gains 2 marks 0.2 / 0.22 = 1 mark 27.06 × 0.9 for 1 mark or 24.354 or 108/27.06 or 0.25055 for 1 mark	2
(ii)	loss in mass due to wear / eroding / corroding / weathering / clipping	ignore damage	1
Total marks			3

Q7.

Question	Answer	Extra information	Marks
	0.054	for 2 marks (0.1 × 13.5)/25 for 1 mark	2
Total marks			2

Q8.

Question	Answer	Extra information	Marks
	0.2288	correct answer gains 3 marks with or without working accept 0.229 or 0.23 if answer is incorrect 28.6 × 0.2 ÷ 1000 (=0.00572) gains 1 mark 0.00572 × 1000 ÷ 25 or ecf gains 1 mark or 28.6 × 0.2 ÷ 25 gains 2 marks	3
Total marks			3