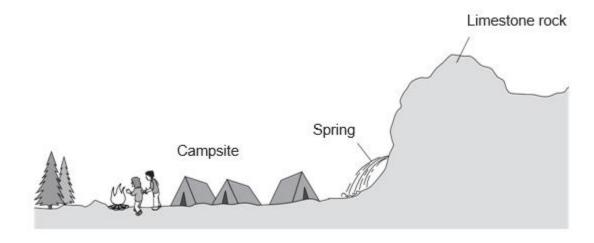
## WATER 1

Q1. This question is about hard water
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(a) A campsite has a spring, where water flows out of limestone rock.



(i) The water from the spring is hard because it contains calcium ions.

How do the calcium ions get into the spring water?

\_\_\_\_\_

\_\_\_\_\_\_

(1 mark)

(ii) A student at the campsite boils some of the spring water in a pan. The inside of the pan becomes coated with a white solid.

What is the white solid?

(1 mark)

(iii) When the student uses soap to wash in the spring water, scum forms. Scum is not formed if the spring water has been boiled and allowed to cool.

Draw a ring around the correct answer to complete the sentence.

The hardness in the spring water is

acidic. permanent.

temporary.

(1 mark)

**(b)** In a laboratory, a student compared the hardness of three different samples of water.

The student measured 20 cm<sup>3</sup> of water into a boiling tube. The student then:

- added a drop of soap solution
- shook the boiling tube for 10 seconds
- looked to see if a permanent lather had formed.

The student repeated the process until a permanent lather formed. The results are shown in the table.

	Number of drops	of soap solution nee	eded to form a per	manent lather
Water sample	Test 1	Test 2	Test 3	Mean
Spring water	13	11	6	
Tap water	7	5	6	6
Distilled water	1	1	1	1

(i)	Calculate the correct mean for spring water.		
		 (2 mark)	

(ii)	Which of the three source	ces of water was hardest?		
	Draw a ring round	the correct answer.		
	distilled water	spring water	tap water	
	Use the results in the table to give a reason for your answer.			
			(	(2 marks)
	ater.		companies add chlorine to st companies adding chlorine	erilise
Q3. (a) drink.	Good quality water is est In the United Kingdom, v		with chlorine to make it saf	(2 marks) e to
Explaii	n why the water is:			

Filtered	
Treated with chlorine	
	(2 marks)

**(b)** Millions of people in Bangladesh drink water from wells that contain high levels of arsenic. Arsenic is poisonous.

The World Health Organisation recommends that there should be no more than 0.01 mg of arsenic per litre in drinking water.

The table gives some information about two instrumental methods of testing for arsenic.

Factor to consider	Laboratory Instrumental Method	Portable Instrumental Method
Cost of equipment	£10 000	£50
Skill level of technician	Highly skilled	Little training needed
Where test is done	Laboratory only	Anywhere
Time to prepare the instrument for the test	5 minutes	10 seconds
Sensitivity of the instrument	0.000001 mg of arsenic per litre of water	0.1mg of arsenic per litre of water

(i)	Use the information in the table to give two advantages and one disadvantage of using
the F	Portable Instrumental Method compared with the Laboratory Instrumental Method.
	(3 marks
(ii)	The information about these two instrumental methods was provided by the
Prof	essional Institute of Water Engineers (PIWE). The Institute has no connection with the
com	panies that make these instruments.
Sugg	gest why many people would accept the views of PIWE rather than the views of the
com	panies that make the instruments.
	(1 mark

**Q4.** The table gives some information about the composition of three samples of water from wells in the Canary Islands, Crete and Cyprus.

	Mineral co	ntent of water in n	ng per litre
lons	Canary Islands	Crete	Cyprus
Calcium, Ca <sup>2+</sup>	28	82	18
Magnesium, Mg <sup>2+</sup>	14	41	13
Sodium, Na <sup>+</sup>	53	7	22
Chloride, Cl	7	143	39
Hydrogen carbonate, HCO <sub>3</sub>	281	5	93
Sulfate, SO <sub>4</sub> <sup>2–</sup>	2	14	16

(a)	Describe and explain now ions get into these samples of water.	
		<del> </del>
		(2 marks)
(b)	The sample of water from Crete is harder than the other two.	
Use	the information in the table to explain why.	
		<del></del>

(	1	mark)	١

(c)	People who use hard water can expect higher costs than people who use	soft water.
Expla	ain why.	
		(2 marks)
(d)	Hard water can be made soft by removing the ions that cause hardness.	,
	e one way these ions can be removed.	
		(1 mark)
<b>Q5.</b> water	Some people use water filters because they are concerned about the quarr.	lity of drinking
(a)	Draw a ring around the correct answer to complete each sentence.	
(i)	One of the active chemicals in many water filters is	
car	bon.	
	gnesium.	
	fur.	
		44 11
		(1 mark)
(ii)	In many areas of the United Kingdom the water is hard.	
The h	nardness in water is caused by ions.	

bromide calcium hydrogen

**Q6.** Read the following information and then answer the questions.

Chlorine - for better, for worse?



(2 marks)

Chlorine is used to make bleaches, plastics and medicines. Swimming pool water is often treated with chlorine. Chlorine is used to make water safe to drink. It is relatively cheap and easy to use. People who drink untreated water risk dying from typhoid and cholera. However, chlorine is a poisonous chemical. It causes breathing difficulties and can kill people. Some people are also allergic to chlorine.

(a)	How does chlorine make water safe to drink?
	(1 mark
(b) contro	The amount of chlorine in swimming pool water should be carefully monitored and blled. Explain why.
	(2 marks
(c) safe to	Developing countries are likely to choose chlorination as their method of making water or drink. Suggest why.
	(1 mark

(d) (i)	A government is setting up an enquiry into the safety of using chlorine.  Suggest why people from all political parties should be represented.
	(1 mark Suggest why the opinion of a well-respected scientist might change the outcome discussion.
	(1 mark
data r	The decision taken about the safety of using chlorine should be based on evidence and ather than on hearsay and opinion. Suggest why.
Q7.	(1 mark In the United Kingdom, water companies supply drinking water to our homes. However,
one m	people are concerned about the taste and quality of the water that is supplied. Describe ethod that people can use at home to improve the taste and quality of the tap water ed. Explain how this method works.

(4 marks
Total marks (38
Total marks (50